

Consider a layer of seawater (heat capacity  $C_p = 4.2 \times 10^3 \text{ Jkg}^{-1}\text{K}^{-1}$ , density  $\rho = 1.027 \times 10^3 \text{ kgm}^{-3}$ ) with a thickness of 1 m and an area of  $10^4 \text{ m}^2$ .

a. Determine how much energy is needed to raise the temperature of the water by  $1^\circ\text{C}$ .

The mass  $m$  of the water layer is  $m = \rho V = 1.027 \times (10^3 \times 1 \times 10^4) = 1.027 \times 10^7 \text{ kg}$ . The energy  $E$  needed to raise the temperature by  $\Delta T$  degrees is given by

$$E = C_p m \Delta T$$

With  $\Delta T = 1^\circ\text{C}$ , we find  $E = 4.31 \times 10^{10} \text{ J} = 43.1 \text{ GJ}$ .

Assume that a constant heat flux  $Q = 400 \text{ Wm}^{-2}$  is applied over the area, for example due to a warmer atmosphere.

b. How long does it take before the temperature of the water layer has increased by  $1^\circ\text{C}$ ?

The mean temperature in the layer of depth  $h$  will change according to the balance

$$\rho h C_p \frac{dT}{dt} = Q$$

From this equation we estimate

$$T(t) \approx T(0) + \frac{Q}{\rho h C_p} t$$

and find with  $T(t_*) - T(0) = 1$  that  $t_* = 1.08 \times 10^4 \text{ s}$ .

During a seasonal cycle, there is heat uptake in an ocean region whereas during the winter the opposite occurs.

c. Assume that the upper 100 m of ocean water is in contact with the atmosphere and that the seasonal temperature change in the water is about  $10^\circ\text{C}$ . Calculate the amount of energy which is stored (released) by the water layer.

For the layer, we now have  $h = 100 \text{ m}$  and hence  $V = 10^6 \text{ m}^3$ . For the seasonal energy change  $E$  we find  $E = \rho V C_p \Delta T = 4.31 \times 10^{13} \text{ J}$ .

d. Determine the same quantity for a land surface and a seasonal temperature change of about  $20^\circ\text{C}$ .

We must estimate a thickness of the layer over which temperature changes occur and a soil density  $\rho_s$ . Dry sand has a density of  $1602 \text{ kgm}^{-3}$  and a heat capacity of  $830 \text{ J(kg K)}^{-1}$  (see [http://www.simetric.co.uk/si\\_materials.htm](http://www.simetric.co.uk/si_materials.htm)). For a layer of 1m of sand, the energy required is  $E = 1602 \times 1 \times 10^4 \times 830 \times 20 = 2.66 \times 10^{11} \text{ J}$ .